NEW DITERPENES, ISOCEMBRENE AND ISOCEMBROL, IN THE OLEORESIN OF PINUS SIBIRICA

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From the neutral part of the oleoresin of Pinus sibirica (R) Mayr, we have isolated two new diterpene compounds with 14-membered rings: a hydrocarbon  $C_{20}\text{H}_{34}\text{O}$ , which we have called isocembrene (I) and isocembrol (II).

Isocembrene with c [ $\alpha$ ] $_D^{20}$  + 60.3° (c 2, 60; chloroform),  $n_D^{20}$  1.5230 contains two double bonds. In the IR spectrum of the compound bands at 890, 1655, and 3085 cm<sup>-1</sup> correspond to an exomethylene double bond, whose presence is confirmed by ozonization, and the band at 970 cm<sup>-1</sup> to a trans-disubstituted and that at 815 cm<sup>-1</sup> to a trisubstituted double bond. The UV spectrum has a maximum at 238 m $\mu$  (log  $\epsilon$  4.3).

The isomerization of I on alumina containing silver nitrate gave cembrene. The exhaustive hydrogenation of I over platinum exide gave octahydrocembrene, and exhaustive hydrogenation over Pd/CaCO<sub>3</sub> gave a hydrocarbon with  $c [\alpha]_D^{20} + 43.0^{\circ}$  (c 0.86; chloroform),  $n_D^{20}$  1.4990, mol. wt. 275 (mass spectrometry) identical with that described by Dauben et al. [1].

The dehydrogenation of isocembrene with selenium gave 2, 9-dimethyl-7-isopropylphenanthrene.

Isocembrol  $C_{20}H_{34}O$  with  $[\alpha]_D^{20} + 80.1^{\circ}$  (c 3.20; chloroform),  $n_D^{20}$  1.5035 is an isomer of cembrol, which we have described previously [2]. We have ascribed bands at 815 and 1665 cm<sup>-1</sup> in the IR spectrum of isocembrol to a trisubstituted double bond, a band at 970 cm<sup>-1</sup> to a trans-disubstituted double bond, and one at 3615 cm<sup>-1</sup> to a tertiary hydroxyl.

Isocembrol contains three double bonds. The UV spectrum shows the absence of conjugation in this compound (in the  $220-400 \text{ m}\mu$  region).

Under the action of acid chlorides, thionyl chloride, and acetic acid, isocembrol readily dehydrates to cembrane, and on the sorbent  $(SiO_2 + AgNO_3)$  to isocembrene (I).

The NMR spectra of the compound do not contradict the structures proposed.

## REFERENCES

- 1. W. G. Dauben, W. E. Thessen, and P. R. Resnick, J. Org. Chem., 30, 1693, 1965.
- 2. A. I. Lisina, A. I. Rezvukhin, and V. A. Pentegova, KhPS [Chemistry of Natural Compounds], 250, 1965.

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